

Erosion behaviour of alumina ceramic coating on mild steel by the modified composition of phosphate binder

by Rendi Reynaldi

General metrics

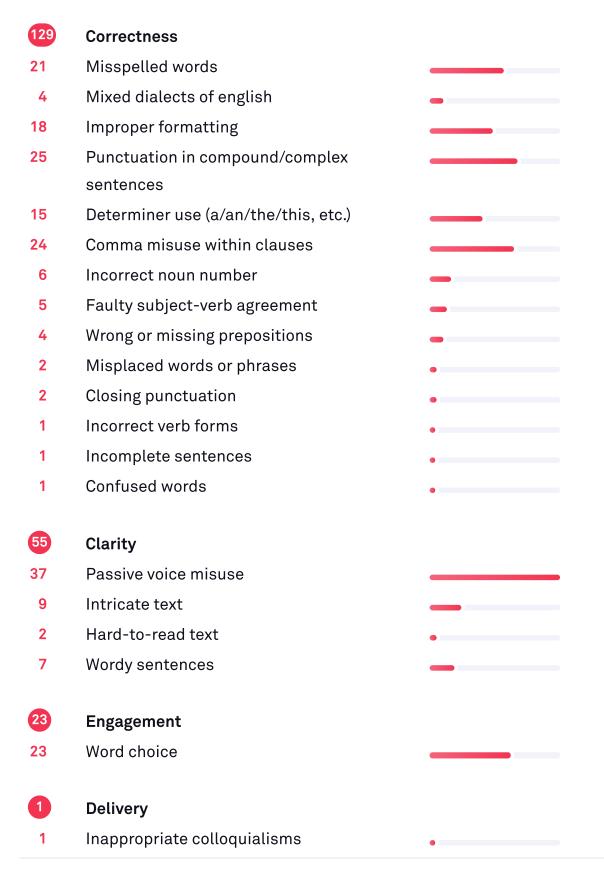
17,742 characters	2,836 words	245 sentences	11 min 20 sec reading time	21 min 48 sec speaking time
Score		Writing I	ssues	
54		208 Issues left	<mark>84</mark> Critical	124 Advanced
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Writing Issues





Unique Words Measures vocabulary diversity by calculating the percentage of words used only once in your document	21% unique words
Rare Words	43%
Measures depth of vocabulary by identifying words that are not among the 5,000 most common English words.	rare words
Word Length	4.4
Measures average word length	characters per word
Sentence Length	11.6
Measures average sentence length	words per sentence



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Erosion <u>behaviour</u>² of alumina ceramic coating on mild steel by the modified composition of phosphate binder

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Abstract. The Pulverizer pipe made of mild steel had erosion failure due to coal dust impacting, thus its service life also reduces. The ceramic coating overlay on the surface of mild steel is one of the appropiate ways to protect the mild steel from erosion. This research is aimed to perform a ceramic coating over the surface of the mild steel using a dipping method to improve its erosion resistance by using the alumina-phosphate ceramic coating. The coating layer is formed by the reaction between monoaluminum phosphate (MAP) as a binder and Al2O3 particles. It transforms into berlinite phase when heated at an elevated temperature. The observation is carried out with the variation of the MAP binder composition Al:P 25:75, 28:72, 30:70 and the Al2O3/MAP slurry is given at 40/60. Scanning electron microscopy is used to characterize the coating morphology. X-ray diffraction is applied to investigate the ceramic coating phases. The gas erosion jet measures erosion resistance of the ceramic coating. From the test result, it can be concluded that the binder composition influenced the erosion behaviour of alumina ceramic coating, the binder with Al:P (30:70) showed the erosion resistance increasing four times compared to the condition without coating.



1. Introduction

The pulverizer pipe is a part of the coal mill pulverizer system which has a function to deliver coal dust to a combustion chamber. Mild steel is commonly used ¹⁹ for the pulverizer pipe material due to its mechanical strength, machinability, weldability, formability and ²⁰ reasonable cost. ²¹Many energy projects also ²² rely on the amount of steel for pipelines and electric power turbine component ²³[1]. Unfortunately²⁴

mild steel does not have good erosion resistance against solid particles such as coal dust. During coal grinding at the working chamber of a coal mill pulverizer, the process releases a certain amount of heat which raises the temperatures to 300°C. Pulverizer pipe and inner cone which made of mild steel is impinged by coal dust particle and exposed to heat continuously, then the erosion takes place. The failure of the pulverizer pipe due to coal dust erosion demontrates in figure 1.

The erosion on ³³ mild steel is particularly induced by repeated plastic deformation of coal dust particle impact. Mbabazi et al ³⁶[2] observed that the erosion rate of mild steel was affected by the impingement angle. The highest erosion rate of mild steel is 6.5 mg.Kg-1 the ³⁷ impingement angle reached an inclination

between 25-30 deg. Ceramic coatings <u>are widely used</u> for the protection of base metal or cement components in chemical, power, and refractory industries <u>against high temperature</u> corrosion and oxidation and for the minimization of wear or erosion [3]. Content from this work may be used ⁴² Attribution 3.0 licence. ⁴³ Any further distribution of this work must maintain attribution to the author(s) and the title of the work, journal citation and ⁴⁴ Published under licence by IOP Publishing Ltd 1

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Figure 1. Erosion failure of pulverizer pipe.

Alumina ceramic coated on steel is used ⁴⁷ to provide better performance and help in decreasing erosion that is caused by pressurized air containing entrained abrasive solid coal particles such as oxides and minerals. When the angle of impingement is less than 30 deg, wear produced is closely similar to abrasion but ⁴⁸ if the angle of impingement is greater than 30 deg to surface, wear produced is released by impact erosion [4]. Ścieszka et al ⁴⁹[5] observed the erosion resistance of mild steel coated-alumina compared to high alumina ceramic by solid ⁵⁰ particle erosion test using pulverized coal dust from a power



plant. The result explained that the erosion resistance of alumina-coated mild steel is four times higher than the high alumina ceramic its self.⁵¹ Troczynski et al [6] have already investigated the behavior of alumina ceramic coating using phosphate binder for the variety of application, ^{53,54} including high- temperature corrosion protection, wear resistance, dielectric properties, non-sticky surface, bioactive ceramic, thermal barrier ceramics, and others. Chen et al ⁵⁵/_[7] expressed in their paper, phosphate inorganic binder plays a role in many applications i.e. ⁵⁶ pairing and joining of material.⁵⁷

This research focused on the use of alumina ceramic coating with phosphate binder on mild steel due to their superior properties and their suitability to the working conditions of pulverizer ⁵⁸ pipe. ⁵⁹ It is

preferred over the sintering method which requires a process at an elevated temperature. The alumina phosphate ceramic coating consists of alumina particle in which aluminum phosphate used as a binder [8]. Aluminum phosphate binder is formed by reacting inorganic mineral or oxide with an acidphosphate solution. It is a sort of inorganic binders used in refractory ceramic coating systems, which has been investigated and applied in the thermal spray coating systems. It is reported that refractories bonded with aluminum phosphate exhibit high strength, high-temperature stability, and abrasion resistance. Wang et al [9] reviewed that phosphate adhesive could be made in low temperature curing with high shear strength and excellent electrical properties. They observed the performance of phosphate adhesive which contains Al(HPO4)2 as the major constituent, has a heat-resistance of 1500°C. In this study, the variety of mole ratio Al/P as a phosphate binder for alumina ceramic that coated mild steel were observed. Furthermore, it was characterized using XRD, SEM-EDS and erosion testing then compared with mild steel without any protection. The results will be recommended for erosion



protection of pulverizer pipe.

2. Experiment Work

2.1. Materials and specimen preparation

The materials used for ceramic coating were alumina (Al2O3) as a filler and Al(OH)3 – H3PO4 as a raw material of binder. Alumina (Al2O3) that has ⁷⁸ a size of $50 - 60 \mu m$ was obtained ⁷⁹ from the local market. All materials that are used ⁸⁰ come up with a technical grade. It is consist ⁸¹ of phosphoric acid (85%), aluminum hydroxide (Al(OH)3), methanol and ⁸² aquades ⁸³ without further purification.

The specimen used in the investigation was mild steel of 5 mm thickness. The mild steel sheet was

cut to a size of 25 mm x 25 mm for ceramic coating. Furthermore, the mild steel specimen was cleaned by methanol to remove scale, corrosion, and other contaminants.

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2.2. Synthesis of binder



The phosphoric acid and aluminum hydroxide were synthesized to achieve monoaluminum phosphate (MAP) binder as the equation (1). $3H3PO4 + Al(OH)3 \rightarrow Al(H2PO4)3 + 3H2O$ (1) The mole ratio of Al/P was varied to find out the influence of the binder composition to the erosion behavior of the coating material as shown in table 1.

Table 1. The mole ratio of Al/P in the synthesis of monoaluminum phosphate binder.

Materials Vario Al/P (25 : 75) us moles of Al(OH) 3: Al/P (28 : 72) H3P04 Al/P (30 : 70) Al(OH)3 12 % 14 % 16 % H3P04 55 % 52 % 51 % Aquades



33 % 34 % 33 %

This mixed solution was reacted ⁹³ at 100° - 120° Celcius to achieve the final product reaction of monoaluminum phosphate. ⁹⁴ Furthermore, the reaction ⁹⁵ was dried ⁹⁶ at the ambient temperature. Figure 2 showed the whole process of this reaction.

Figure 2. Synthesis of monoaluminum phosphate.

2.3. Synthesis of alumina phosphate ceramic coating The alumina phosphate ceramic coating was synthesized from aluminum oxide particle and monoaluminum phosphate binder with 40:60 of the ratio [3] and applied to the mild steel which ⁹⁷ has already cleaned with methanol to remove the scale and grease, by dipping technique at the constant time. To gain the hardness of the aluminum coating⁹⁸, it was dried at the various temperature and illustrated in Figure 3.

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Figure 3. Heat treatment of the aluminum-phosphate ceramic coating.

2.4. Characterization
Several tools were used to characterize the coating. Optical microscope
(OLYMPUS SZ61) was used to
figure the coating surface, Scanning Electron Microscope (Hitachi SU3500 at 5kV and 40% spot intensity) to observe the microstructure of the coating and interface bonding between steel and coating. Energy Dispersive Spectrometer
(EDS) was applied to check the alumina particle composition as depicted in figure 4. It showed the alumina was high purity.

Figure 4. Elemental analysis of alumina by EDS.

The density of the coating <u>was counted</u> ¹⁰⁶ by Pycnometer (Pyrex, 5 mL). X-Ray Diffraction (SmartLab X-Ray Diffractometer) was used to characterize the binder and to operate at 30 kV and 30 mA scanning at 10° (20) to 60°. Erosion tester (TR-470) used to characterize the erosion properties of the coating. This test according to ASTM G76-02 (standard test method for conducting erosion test by solid particle impingement using gas jets). After the test piece being placed at a jig, it will undergo impact load of alumina particle from the nozzle with speed of 30 m/s until 70 m/s, 90° angle and particle feed rate 2 gr/min for 10 minutes.

3. Result and Analysis

3.1. Monoaluminum phosphate binder

He <u>et al</u>¹¹³ [3], Emmerson <u>et al</u>¹¹⁴ and Wagh [10], reviewed that aluminum hydrogen phosphate or Al(H2PO4)3 is the phase <u>being expected</u>¹¹⁶ to be present on this binder which can be reacted with the aluminum oxide particle to form berlinite. The XRD result can be seen ¹¹⁷ figure 5.

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The peak shows other compound aluminum hydrogen phosphate hydrate Al(H2PO4) nH2O. This phase was detected because the binder aluminum still consists of water, due to the heating process before. The XRD result can be the basis of this research to continue the synthesized between binder and the aluminum oxide particles to achieve coating material. Aluminum hydrogen

phosphate hydrate was formed as much as aluminum hydrogen phosphate based on their height of intensity peaks.

Figure 5. XRD result of synthetic binder.

3.2. Characterizations of alumina-phosphate ceramic coating Observation of the coating surface was conducted to find the agglomeration of the particle, crack, and the impurity from other material. ¹²³ This observation is important explained by Colonetti ¹²⁶ research [11]. Colonetti ¹²⁷ reviewed that crack and porosity influence to the mechanical properties of the coating.

(a) (b) (c)

Figure 6. SEM micrograph of surface morphology of the coating affected by mole ratio of Al/P

a. Al/P (25:75) b. Al/P (28 :72) c. Al/P (30:70).

As seen in figure 6, the surfaces of the coating has a uniform particle distribution, leak and fissures are observed ¹³² in this coating. The mole ratio of Al/P leads the binder ability to form the compacted ceramic coating. The more the mole ratio of Al in Al/P the denser of the coating becomes. Wagh <u>et al</u> $[10]^{136}$ suggest that the bonding phase is <u>a converted</u> from <u>a very small</u> amount of alumina.

Al2O3 particles disperse in the Al2O3-berlinite system and the product is a thin coating of berlinite on the alumina. As depicted in figure 6 (a-c), the increase of ¹ P ratio (as H3PO4) creates the amount of porosity due to gas formation when the dipping process. This phenomenon relates with the increase of ¹⁴¹MAP acidity which ¹⁴³will react with mild steel surface and then releases hydrogen gas.

Figure 7. XRD result of alumina-phosphate ceramic coating contained Al/P binder (28:72).

XRD test was used to characterize the compound of the coating with binder ratio Al/P (28:72). Figure

7 shows the main compound of this coating are aluminum oxide, cristobalite and little amount of

berlinite in 20°-30°. ¹⁵¹Berlinite is described as a small peak in a diagram, ¹⁵²since it serves only as a binder in the material. He et al [3], and Colonetti et al [11] also

detected berlinite at 20° - 30° with a small peak.

3.3. Effect of density to an alumina-phosphate ceramic coating Characterization of coating morphology was conducted to prove the effectiveness of the binder to the alumina ceramic coating. Wagh et.al ¹⁵⁷[10] observed the effect of binder density on the distance and compactness of the coating ¹⁵⁸ ¹⁵⁹. Table 2 shows the various density of the binder with a different mole ratio of Al/P.

Table 2. The density of the alumina-phosphate ceramic coating.

Density (g/cm3) Al/P (25:75) 1.6704 Al/P (28 :72) 1.7029 Al/P (30:70) 1.7158

a Binder

□b Binder

□c Binder

Void



Particl		
Void		
Particl		
∏Void		

Particl

Figure 8. SEM micrograph a. Al/P (25:75) , b. Al/P (28:72) and c. Al/P (30:70).

Figure 8 shows in addition to the effect of the density, the void between the particle can also <u>be caused</u> by the distribution and size of the Al2O3 particle [10]. Void on aluminum ceramic coating was filled by the berlinite as synthesized between the binder and the Al2O3 particle. The reaction product will fill the space of the particle and make it compact. This condition <u>occurrs</u> when the product reaction has less density than the Al2O3 particle. Figure 8 (c) shows the product reaction of the binder with molar

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ratio ¹⁶⁹ Al/P (30:70) has less density than Al2O3 particle and make ¹⁷⁰ the coating ¹⁷¹ more compact compared to other molar ratios. ¹⁷¹ The compactness of the coating ¹ is important ¹⁷³ which ¹⁷⁴ related to the mechanical properties of the coating. ¹⁷⁵

3.4. Erosion behavior of the alumina-phosphate ceramic coating Figure 9 shows the result of erosion test, as described in the diagram, it can be seen that coating with the molar ratio of Al/P (30:70) has 5mg/Kg loss material, compare ¹⁷⁶ to the steel without any protection that has only 20 mg/Kg loss material. **G** grammarly

uncoated
Erosion rate (mg/Kg)
20
15
<u>25 :</u> 75
5
П
28 : 72
<u>30 :</u> ¹⁷⁹ 70



0

Mole ratio of Al : P

Figure 9. Erosion testing of alumina-phosphate ceramic coating and <u>uncoated</u> mild steel.

The result indicates that the erosion resistance of the coating steel is four-time than steel without any protection. This molar ratio also has a higher result than other ratios, it might be caused by the compactness of the coating and the void effect between the particle as the SEM test shows before. He et al [3] reviewed the abrasion resistance of the coating ¹⁸⁶ on their research which ¹⁸⁷ has two-time wear durability compared to steel without coating. The microstructure of the coating ¹⁸⁹ after erosion ¹⁸⁹ test can be seen ¹⁹⁰ figure 9.

a b Micro Crack c

Micro Crack

Micro Crack



Creater Micro Crack

Micro Crack

Figure 10. SEM micrograph on magnification x850 after erosion testing a. Al/P (25:75) , b. Al/P (28:72) dan c. Al/P (30:70).

The microcrack was found ¹⁹² on the coating after the erosion test, in which the crack can develop due to the impact from erosion particle and also an indication of mechanical properties of the binder. Mol ratio Al/P (30:70) has a less amount of crack, and it is related to the erosion test result of this mol ratio

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which exibit less weight loss during the test compared to other ratios. ¹⁹⁷ A crater was observed ¹⁹⁸ in the mol ratio Al/P (28:72) this crater can happen through the test.

4. Conclusion

The mole ratio of Al/P in MAP binder influence the properties and erosion behavior of alumina ceramic coating. The aluminum phosphate ceramic coating was successfully coated to the steel and increase erosion resistance of the steel four times than steel without any protection. The best mole ratio of this aluminum phosphate ceramic coating was obtained in Al/P (30:70) and heated at 220°C to form

aluminum phosphate or berlinite as the binding phase of this coating in this research.

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research grant. Authors also would say thanks to POLMAN Bandung an all colleagues for the support in all sector.



1.	<mark>doi</mark> → DOI	Misspelled Words	Correctness
2.	behaviour → behavior	Mixed Dialects of English	Correctness
3.	E-mail :	Improper Formatting	Correctness
4.	, thus → . Thus, ; thus	Punctuation in Compound/Complex Sentences	Correctness
5.	appropiato → appropriate	Misspelled Words	Correctness
6.	is aimed	Passive Voice Misuse	Clarity
7.		Intricate Text	Clarity
8.	is formed	Passive Voice Misuse	Clarity
9.	the berlinite	Determiner Use (a/an/the/this, etc.)	Correctness
10.	is carried	Passive Voice Misuse	Clarity
11.	, and	Punctuation in Compound/Complex Sentences	Correctness
12.	is given	Passive Voice Misuse	Clarity
13.	is used	Passive Voice Misuse	Clarity
14.	be concluded	Passive Voice Misuse	Clarity
15.	behaviour → behavior	Mixed Dialects of English	Correctness
16.	, the $→$. The	Hard-to-read text	Clarity
17.	coating → coat	Word Choice	Engagement
18.	, which	Punctuation in Compound/Complex	Correctness



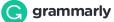
		Sentences	
19. is commonly	used	Passive Voice Misuse	Clarity
20. , and		Comma Misuse within Clauses	Correctness
pulverizer pip mechanical s	commonly used for the be material due to its strength, machinability, formability and reasonable	Hard-to-read text	Clarity
22. <mark>also</mark>		Wordy Sentences	Clarity
23. component -	→ components	Incorrect Noun Number	Correctness
24. Unfortunate	ly,	Comma Misuse within Clauses	Correctness
25. good → excel	llent	Word Choice	Engagement
26. the coal		Determiner Use (a/an/the/this, etc.)	Correctness
27. , which		Punctuation in Compound/Complex Sentences	Correctness
28. , which		Punctuation in Compound/Complex Sentences	Correctness
29. is →are		Faulty Subject-Verb Agreement	Correctness
30. particle → pa	articles	Incorrect Noun Number	Correctness
31. ${, \text{then}} \rightarrow ; \text{the}$	en, , and then, . Then	Punctuation in Compound/Complex Sentences	Correctness
32. demontrates	G → demonstrates, demonstrated	Misspelled Words	Correctness

G grammarly

33.	on → of	Wrong or Missing Prepositions	Correctness
34.	particularly → mainly	Word Choice	Engagement
35.	is particularly induced	Passive Voice Misuse	Clarity
36.	et al → et al.	Comma Misuse within Clauses	Correctness
37.	, the	Punctuation in Compound/Complex Sentences	Correctness
38.	are widely used	Passive Voice Misuse	Clarity
39.	against high → against high	Improper Formatting	Correctness
40.	high-temperature	Misspelled Words	Correctness
41.	minimizes	Wordy Sentences	Clarity
42.	be used	Passive Voice Misuse	Clarity
43.	licence → license	Mixed Dialects of English	Correctness
44.	, and	Comma Misuse within Clauses	Correctness
45.	<mark>licence</mark> → license	Mixed Dialects of English	Correctness
46.	<mark>doi</mark> → DOI	Misspelled Words	Correctness
47.	is used	Passive Voice Misuse	Clarity
48.	, but	Punctuation in Compound/Complex Sentences	Correctness
49.	ot al → et al.	Comma Misuse within Clauses	Correctness
50.	<mark>solid</mark> → robust, substantial, reliable	Word Choice	Engagement
51.	its solf → itself	Wordy Sentences	Clarity



52.	et al → et al.	Comma Misuse within Clauses	Correctness
53.	various applications	Wordy Sentences	Clarity
54.	application → applications	Incorrect Noun Number	Correctness
55.	et al → et al.	Comma Misuse within Clauses	Correctness
56.	i.e.,	Comma Misuse within Clauses	Correctness
57.		Intricate Text	Clarity
58.	the pulverizer	Determiner Use (a/an/the/this, etc.)	Correctness
59.		Intricate Text	Clarity
60.	is preferred	Passive Voice Misuse	Clarity
61.	, which	Punctuation in Compound/Complex Sentences	Correctness
62.	particle → particles	Incorrect Noun Number	Correctness
63.	is reported	Passive Voice Misuse	Clarity
64.	et al → et al.	Comma Misuse within Clauses	Correctness
65.	, which	Punctuation in Compound/Complex Sentences	Correctness
66.	major → principal	Word Choice	Engagement
67.	were observed	Passive Voice Misuse	Clarity
68.	, and	Comma Misuse within Clauses	Correctness
69.	then	Wordy Sentences	Clarity
70.	compared with → compared with	Improper Formatting	Correctness

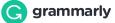


71.	with mild → with mild	Improper Formatting	Correctness
72.	mild steel → mild steel	Improper Formatting	Correctness
73.	steel without → steel without	Improper Formatting	Correctness
74.	without any → without any	Improper Formatting	Correctness
75.	any protection → any protection	Improper Formatting	Correctness
76.	be recommended	Passive Voice Misuse	Clarity
77.	the pulverizer	Determiner Use (a/an/the/this, etc.)	Correctness
78.	that has → with	Wordy Sentences	Clarity
79.	was obtained	Passive Voice Misuse	Clarity
80.	are used	Passive Voice Misuse	Clarity
81.	consist → consists	Faulty Subject-Verb Agreement	Correctness
82.	, and	Comma Misuse within Clauses	Correctness
83.	<mark>aquades</mark> → equates	Misspelled Words	Correctness
84.	was cut	Passive Voice Misuse	Clarity
85.	mild → calm, soft, sweet	Word Choice	Engagement
86.	doi → DOI	Misspelled Words	Correctness
87.	were synthesized	Passive Voice Misuse	Clarity
88.	the behavior	Determiner Use (a/an/the/this, etc.)	Correctness
89.	, as	Punctuation in Compound/Complex	Correctness

		Sentences	
90.	25 :	Improper Formatting	Correctness
91.	28 :	Improper Formatting	Correctness
92.	30 :	Improper Formatting	Correctness
93.	was reacted	Passive Voice Misuse	Clarity
94.		Intricate Text	Clarity
95.	reaction → result, response	Word Choice	Engagement
96.	was dried	Passive Voice Misuse	Clarity
97.	, which	Punctuation in Compound/Complex Sentences	Correctness
98.	To gain the hardness of the aluminum coating	Misplaced Words or Phrases	Correctness
99.	<mark>doi</mark> → DOI	Misspelled Words	Correctness
100.	were used	Passive Voice Misuse	Clarity
101.	was used	Passive Voice Misuse	Clarity
102.	, to	Punctuation in Compound/Complex Sentences	Correctness
103.	coating → sheet, cladding, layer	Word Choice	Engagement
104.	composition as → composition as	Improper Formatting	Correctness
105.	, as	Punctuation in Compound/Complex Sentences	Correctness



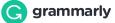
106.	was counted	Passive Voice Misuse	Clarity
107.	, according	Punctuation in Compound/Complex Sentences	Correctness
108.	<mark>solid</mark> → substantial	Word Choice	Engagement
109.	being placed	Passive Voice Misuse	Clarity
110.	an impact	Determiner Use (a/an/the/this, etc.)	Correctness
111.	the speed, or a speed	Determiner Use (a/an/the/this, etc.)	Correctness
112.	, and	Comma Misuse within Clauses	Correctness
113.	et al → et al.	Comma Misuse within Clauses	Correctness
114.	et al → et al.	Comma Misuse within Clauses	Correctness
115.],	Comma Misuse within Clauses	Correctness
116.	being expected	Passive Voice Misuse	Clarity
117.	be seen	Passive Voice Misuse	Clarity
118.	<mark>doi</mark> → DOI	Misspelled Words	Correctness
119.	compound aluminum hydrogen phosphate hydrate	Intricate Text	Clarity
120.	was formed	Passive Voice Misuse	Clarity
121.	the synthetic, or a synthetic	Determiner Use (a/an/the/this, etc.)	Correctness
122.	was conducted	Passive Voice Misuse	Clarity
123.		Intricate Text	Clarity



124.	important → essential	Word Choice	Engagement
125.	important,	Punctuation in Compound/Complex Sentences	Correctness
126.	<mark>Colonetti</mark> → Colonetti's	Incorrect Noun Number	Correctness
127.	<mark>Colonetti</mark> → Coletti	Misspelled Words	Correctness
128.	to → on	Wrong or Missing Prepositions	Correctness
129.	the mole	Determiner Use (a/an/the/this, etc.)	Correctness
130.	has → have	Faulty Subject-Verb Agreement	Correctness
131.	, and	Comma Misuse within Clauses	Correctness
132.	are observed	Passive Voice Misuse	Clarity
133.	$\frac{\text{coating}}{\text{coating}} \rightarrow \text{layer}$	Word Choice	Engagement
134.	, the	Punctuation in Compound/Complex Sentences	Correctness
135.	$\frac{\text{coating}}{\text{coating}}$ > surface, layer, sheet	Word Choice	Engagement
136.	ot al → et al.	Comma Misuse within Clauses	Correctness
137.	-a converted	Determiner Use (a/an/the/this, etc.)	Correctness
138.	<mark>a very small</mark> → a minimal, a tiny	Word Choice	Engagement
139.	, and	Punctuation in Compound/Complex Sentences	Correctness
140.	increase of	Wordy Sentences	Clarity



141.	with \rightarrow to	Wrong or Missing Prepositions	Correctness
142.	increase of → rise in, rise of	Word Choice	Engagement
143.	, which	Punctuation in Compound/Complex Sentences	Correctness
144.	the mild	Determiner Use (a/an/the/this, etc.)	Correctness
145.	<mark>doi</mark> → DOI	Misspelled Words	Correctness
146.	was used	Passive Voice Misuse	Clarity
147.	<mark>7</mark> → Seven	Improper Formatting	Correctness
148.	compound → mixture, synthesis	Word Choice	Engagement
149.	aro → is	Faulty Subject-Verb Agreement	Correctness
150.	, and	Comma Misuse within Clauses	Correctness
151.		Intricate Text	Clarity
152.	diagram,	Punctuation in Compound/Complex Sentences	Correctness
153.	ot al → et al.	Comma Misuse within Clauses	Correctness
154.],	Punctuation in Compound/Complex Sentences	Correctness
155.	<mark>et al</mark> → et al.	Comma Misuse within Clauses	Correctness
156.	was conducted	Passive Voice Misuse	Clarity
157.	et.al → et al.	Comma Misuse within Clauses	Correctness



ceasting → surface, sheet, layer, coatWord ChoiceEngagementIntricate TextClaritydeneity → thickness, frequencyWord ChoiceEngagementParticl → Particle, PartialMisspelled WordsCorrectnessParticl → Particle, PartialMisspelled WordsCorrectnessParticl → Particle, PartialMisspelled WordsCorrectnessParticl → Particle, PartialMisspelled WordsCorrectnessbe causedPassive Voice MisuseClarityac cynthesized → as-synthesizedMisspelled WordsCorrectnessmolar.Closing PunctuationCorrectnessmolar.Closing PunctuationCorrectnessratio → RatioImproper FormattingCorrectnessmake → makesFaulty Subject-Verb AgreementCorrectnessimportant → essentialWord ChoiceEngagementimportant → essentialWord ChoiceEngagementimportant → essentialWord ChoiceEngagementwhichPunctuation in CorrectnessCorrectnessceasting → surface, sheetWord ChoiceEngagementcompare → comparedIncorrect Verb FormsCorrectness25 :Improper FormattingCorrectness			
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compare → compared Incorrect Verb Forms Correctness	, which	Compound/Complex	Correctness
	coating → surface, sheet	Word Choice	Engagement
25: Improper Formatting Correctness	compare → compared	Incorrect Verb Forms	Correctness
	25 :	Improper Formatting	Correctness



178.	28 :	Improper Formatting	Correctness
179.	30 :	Improper Formatting	Correctness
180.	The mole, or A mole	Determiner Use (a/an/the/this, etc.)	Correctness
181.	uncoated mild → mild uncoated	Misplaced Words or Phrases	Correctness
182.	ratios → rates	Word Choice	Engagement
183.	, it → ; it, , and it, . It	Punctuation in Compound/Complex Sentences	Correctness
184.	be caused	Passive Voice Misuse	Clarity
185.	ot al → et al.	Comma Misuse within Clauses	Correctness
186.	coating → surface, layer	Word Choice	Engagement
187.	, which	Punctuation in Compound/Complex Sentences	Correctness
188.	$\frac{coating}{coating}$ \rightarrow surface, sheet	Word Choice	Engagement
189.	the erosion	Determiner Use (a/an/the/this, etc.)	Correctness
190.	be seen	Passive Voice Misuse	Clarity
191.	Creater → Creator, Create	Misspelled Words	Correctness
192.	was found	Passive Voice Misuse	Clarity
193.	ratio.	Closing Punctuation	Correctness
194.	<mark>doi</mark> → DOI	Misspelled Words	Correctness



195.	which → Which	Improper Formatting	Correctness
196.	exibit → exhibit	Misspelled Words	Correctness
197.	which exibit less weight loss during the test compared to other ratios.	Incomplete Sentences	Correctness
198.	was observed	Passive Voice Misuse	Clarity
199.	the MAP	Determiner Use (a/an/the/this, etc.)	Correctness
200.	For	Inappropriate Colloquialisms	Delivery
201.	jiang → Jiang	Misspelled Words	Correctness
202.	, and	Comma Misuse within Clauses	Correctness
203.	, and	Comma Misuse within Clauses	Correctness
204.	to	Wrong or Missing Prepositions	Correctness
205.	, which	Punctuation in Compound/Complex Sentences	Correctness
206.	the HIBAH	Determiner Use (a/an/the/this, etc.)	Correctness
207.	POLMAN → Polman	Confused Words	Correctness
208.	sector → sectors	Incorrect Noun Number	Correctness